

1 **Impact of anti-ageing compounds on oxidation ageing kinetics of bitumen by**
2 **infrared spectroscopy analysis¹**

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37 Abstract

38 This paper investigates the effect of different anti-ageing compounds (AACs) on the oxidation
39 kinetics of bitumen using Fourier Transformation Infrared (FTIR) spectroscopy. Twenty different
40 AACs were examined, including new and existing AACs for bitumen and polymer products. The
41 AACs were mixed with bitumen to fabricate thin film samples of AAC-modified bitumen which
42 were subjected to laboratory oven ageing at 100 °C with different ageing periods up to 504 hours.
43 A Normalized Carbonyl Index (NCI) was proposed based on a selected reference peak (1377 cm^{-1})
44 to eliminate the impact of the inherent carbonyl content from the bitumen or AACs and manifest
45 the carbonyl growth rate for evaluating the AACs' anti-ageing performance. It was found the
46 activation energy of fast-term oxidation can be utilized to quantitatively screen the anti-ageing
47 compounds and evaluate their anti-ageing effectiveness in terms of decreasing the formation of
48 carbonyl groups in bitumen. AACs that exhibited high anti-ageing performance were those
49 contained furfural, Irganox acid with sodium montmorillonite, furfural with DLTDP, and high
50 concentrations (e.g., 15%) of Irganox acid. The proposed protocol should be followed by further
51 laboratory rheological and mechanical tests on the AAC-modified bitumen with different binder
52 sources.

53

54 *Keywords:* Ageing of bitumen, Anti-ageing compounds, Oxidation kinetics, FTIR, Oven ageing,
55 Carbonyl index

56 1. Introduction

57 Ageing of bitumen is identified by the increase in its stiffness which causes adverse
58 changes to the mechanical performance of roads and eventually leads to ageing failure in a form
59 of thermal or fatigue cracking. Oxidative ageing of the bitumen is typically quantified by the
60 development of carbonyl functional groups using Fourier Transformation Infrared Spectroscopy
61 (FTIR) test [1]. This is fundamentally due to the fact that a linear relationship exists between the
62 carbonyl content formed and the oxygen absorbed by the bitumen. A correlation also exists
63 between the bitumen's hardening susceptibility and the carbonyl content [2]. Wide range of FTIR
64 applications for bitumen exist. FTIR applications include materials and ageing recognition.
65 Additionally, researchers utilized FTIR to examine the interaction between the bitumen and
66 modifiers and the moisture effects on ageing [3-5]. A detailed explanation for these functions is
67 listed in the literature [6].

68 For ageing recognition and quantification, researchers have established different
69 parameters to characterise the carbonyl-growth, one of which is carbonyl area (*CA*) that is defined
70 as the integral area under the absorbance curve from FTIR tests within a wavelength range from
71 1820 to 1650 cm⁻¹ [7]. The carbonyl area was intensively used to quantify the ageing performance
72 of the bitumen; however, it has limitations in comparing the modified binders when the additives
73 themselves contain carbonyl functional groups in their primary chemical form and may change
74 during the ageing process [5]. Additionally, the fixed wavelength range (1820 to 1650 cm⁻¹) can
75 cause inaccuracy in the calculation of the carbonyl area due to the changes encountered in the
76 absorbance wave pattern within that range. This case is more pronounced when bitumen lacks
77 carbonyl prior to ageing. This can lead to a negative value in the carbonyl area, as shown in the
78 following sections. Herrington (2012) measured the normalized carbonyl area in a range from 1640
79 to 1810 cm⁻¹ and divided it by the area under 1600 cm⁻¹ peak (using 1810 cm⁻¹ as a baseline), to
80 solve the variations in sample concentration [8]. Liu et al. (2015) introduced a carbonyl index as
81 the ratio of carbonyl area under 1700 cm⁻¹ peak to that of methylene group under 1375 cm⁻¹ peak
82 to eliminate the effect of bitumen sample thickness on the carbonyl area [9]. Another adopted
83 method was to divide the carbonyl area under the 1700 cm⁻¹ peak by the summation of all
84 absorbance areas for the bitumen [10] or a portion of it [11]. Similarly, the sulfoxide index (which
85 is an additional oxidation measure), was calculated by the same procedures but under a peak of
86 1031 cm⁻¹. It can be seen that there is no agreed or consistent method to determine the carbonyl
87 content and the decision was rather arbitrary depending on individual choices.

88 The attempt to characterize the ageing quantity of modified bitumen binders such as SBS
89 polymer modified binders by carbonyl formation is another complication. This led to adopting a
90 new set of testing methods and conditions [12]. Few attempts were made to address this issue [4].
91 Zhao et al. (2010) studied the ageing characteristics and materials interaction of polymer modified
92 bitumen using two carbonyl parameters [13]. One was using carbonyl index to study ageing
93 properties of the bitumen-polymer blend. The second parameter was carbonyl area to address the
94 changes occurring in polymer and base bitumen individually. However, no justification was
95 provided for using those two parameters. Therefore, it is difficult to compare the ageing resistance
96 for the bitumen with different modifiers subjectively. Furthermore, it is unclear which FTIR
97 parameter is effective to rank the anti-ageing compounds' (AACs) performance in reducing ageing
98 of the bitumen. Reasons for this attributed to the adoption of various ageing conditions (such as

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short-term ageing, long-term ageing, different temperatures and pressure conditions), evaluation criteria (rheological properties or chemical changes), and parameter used (ageing indices).

Many AACs were tested to enhance the bitumen's mechanical performance and long-term durability, and the work in this area also continues due to the persistent new findings and AACs developed. While extensive work is available to model the viscoelastic asphalt mixtures and aggregate structures [14, 15], the ageing characterization the bitumen binders modified by AACs are needed in order to develop a comprehensive multi-physical model for the durability prediction of the asphalt mixtures in the field. Furthermore, linking the mechanical performance with the oxidation kinetics will produce a reliable integrated pavement performance model. Thus, a consistent primary selection criterion is very needed for screening these anti-ageing compounds and the effect of the AACs on the ageing kinetics of bitumen requires further studies, particularly for those AACs with inherent carbonyl functional groups.

In summary, the purpose of this paper is to develop a unified AAC screening method and consistent ageing parameters to evaluate the anti-ageing performance of different types of AACs and comparing the ageing kinetics of the bitumen modified by those AACs.

2. Materials and Testing Methods

A type of bitumen, classed as a 40/100 according to BSEN14023:2010 for general asphalt applications and road constructions, was used as a base (control) binder to examine the effect of adding AACs, where Table 1 lists its engineering specifications. The reason behind examining a single type of bitumen was to withdraw binder-source effects on the AAC results. Others have used a similar approach to neglect the binder type variances on ageing kinetics [9, 11, 16 and 17].

TABLE 1 Conventional properties of the control binder

Property	Value
Penetration @ 25 °C (0.1 mm)	45-80
Softening Point (°C)	>45
Flash Point (°C)	>250
Force Ductility @ 5 °C (J/cm ²)	>3

Nomination of AACs was based on four aspects: 1) materials that proved effective in previous researches but require further investigations, 2) new polymer modifiers that have not been tested on bitumen binders before, 3) different multifunctional nanomaterials (materials in nanoscale size range in at least one dimension, that could work as exfoliators/intercalators, hydroxyl groups providers, metal fillers, etc...), and 4) some common bitumen AACs that were added for comparison. The selected AAC compounds vary in their anti-oxidation functionality, structure and chemical behaviour, this is to ensure the study covers a wide range of anti-ageing materials. The concentrations of AACs were based on two aspects, namely, the availability of materials to be used in large quantities, and the previous literature available concerning each additive's expected reaction. Some additives were tested at different concentrations. A list of AACs and their concentrations are summarized in Table 2. Details of each additive are provided below.

Furfural (2-Furaldehyde) with (Dilauryl) Thiodipropionate (DLTDP) (didodecyl 3, 3'-thiopropionate) compound was investigated by Apeagyei (2011) in terms of the rheological properties of the binder. A percentage of 3.5% (4:3 furfural: DLTDP) achieved 40% reduction in

138 the ageing hardening of modified binders compared with unmodified binders. However, the
 139 chemical properties and ageing kinetics were not investigated [14].

140 No researches were found on the use of Irganox 1076 (Octadecyl-3-(3,5-di-tert. butyl-4-
 141 hydroxyphenyl)-propionate) for bitumen. Irganox 1076 is a commercial polymer additive with low
 142 volatility, good resistance to extraction and thermo-oxidative degradation (the process of oxygen-
 143 containing groups formation that occurs under the comprehensive effects of light, heat and
 144 oxygen).

145 In addition, two other types of Irganox additives were tested, MD1024 and Irganox acid.
 146 MD 1024 (Benzene propanoic acid, 3,5-bis(1,1-dimethylethyl)-4-hydroxy-, 2-[3-[3,5-bis(1,1-
 147 dimethylethyl)-4-hydroxyphenyl]-1-oxopropyl] hydrazide) is used for protection against metals
 148 and minerals contamination at high temperatures. Whereas, Irganox acid is a hindered phenol
 149 organic (3,5-di-tert-butyl-4-hydroxyphenylpropionic acid) that has been synthesized at Aston
 150 University Chemical Engineering laboratories and currently being tested as a polymer antioxidant.

151 Hydrated lime has been included for comparison purposes in the testing program as it is a
 152 common AAC for bitumen [19-23]. It is known for reducing the rate of age hardening. This
 153 reduction is attributed to the absorption by lime of asphalt components that otherwise would have
 154 increased the sensitivity of the asphalt to the oxidation products [22]. The use of emulsifiers such
 155 as lime is a common practice especially in cold recycled asphalt pavements [24]. The end result of
 156 using lime emulsifier is hydrated lime. Therefore, it has been included as one of the AAC in this
 157 study.

158 Nanomaterials such as Cloisite C20A, bentonite HCT and sodium montmorillonite were
 159 included in the testing program. Ghli (2006), Ortega et al. (2017), Kumar and Suman (2017),
 160 Hassan et al., (2012) and Yao et al., (2013) reported the beneficial effects of adding nanomaterials
 161 to bitumen. The benefits of those AACs include the increase of bitumen's thermal stability,
 162 resistance against permanent deformations and strengthening of bitumen's bonding with rubber
 163 modifiers. Furthermore, these AACs are believed to decrease the oxygen diffusion during ageing.
 164 However, debate exists on whether these improvements can justify their applications on wider
 165 range conditions in the field [25-29].

166 Other additives were also tested, such as tetramethyl thiuram disulphide, TRIS and
 167 trimethylolpropane. tetramethyl thiuram disulphide is used as a polymerization initiator and
 168 accelerator in rubber industry [30]. TRIS (hydroxymethyl-aminomethane), according to Wilkes
 169 and Davies (2010)) can be used as a dispersant for asphaltene inhibition of hydrocarbon fluids
 170 such as bitumen and crude oil [31]. Whereas, trimethylolpropane is a stable compound under
 171 different environmental and light exposure conditions and was included due to its ability to reduce
 172 oxidation of trimethylolpropane esters based on palm oil and palm kernel oils [32]. The technical
 173 and safety properties of these materials are listed in the manufacturers' technical and safety
 174 datasheets. Dealing with furfural is considered hazardous to human health, which requires special
 175 measures during the mixing processes in the field outside the laboratory-controlled conditions.

176 177 TABLE 2 Anti-ageing compounds and concentrations.

Additive	The concentration of additive added (% by mass of bitumen binder)
Irganox acid	5, 10, 15
Tetramethylthiuram disulphide	1

Cloisite C20A	10
Bentonite HCT	10
Sodium montmorillonite	10, 15
TRIS (hydroxymethyl-aminomethane)	5, 10
Calcium hydroxide	10
(3:1:1) TRIS (hydroxymethyl-aminomethane): bentonite: calcium hydroxide	10
Irganox 1076	15
MD1024	0.1
(3:2) sodium montmorillonite: Irganox acid	25
(1:1) calcium hydroxide: TRIS (hydroxymethyl-aminomethane)	20
(4:3) furfural: DLTDP ((Dilauryl) Thiodipropionate)	3.5
Furfural	2
(1:5) furfural: Irganox 1076	12
Trimethylolpropane TMP	10

178

Testing samples were prepared by mixing the additives with bitumen (mass of bitumen samples 1-3 g) using solvent blending method, i.e., mother liquor melting (Li et al (2017)) to achieve better dispersion of the AACs into bitumen, as well as, due to the nature of FTIR testing which requires placing a thin-film sample on a sodium chloride testing plate [33]. Dichloromethane was used as a solvent for its high bitumen dissolving ability at low temperatures such as at room temperature, high evaporation rate and low viscosity [10], while others used solvents such as kerosene due to its low cost and availability [33].

Samples were prepared by mixing additives, bitumen and equal concentrations of dichloromethane by means of an ultrasonic shaker for at least 30 minutes to ensure complete dispersion at room ambient temperature (20 °C).

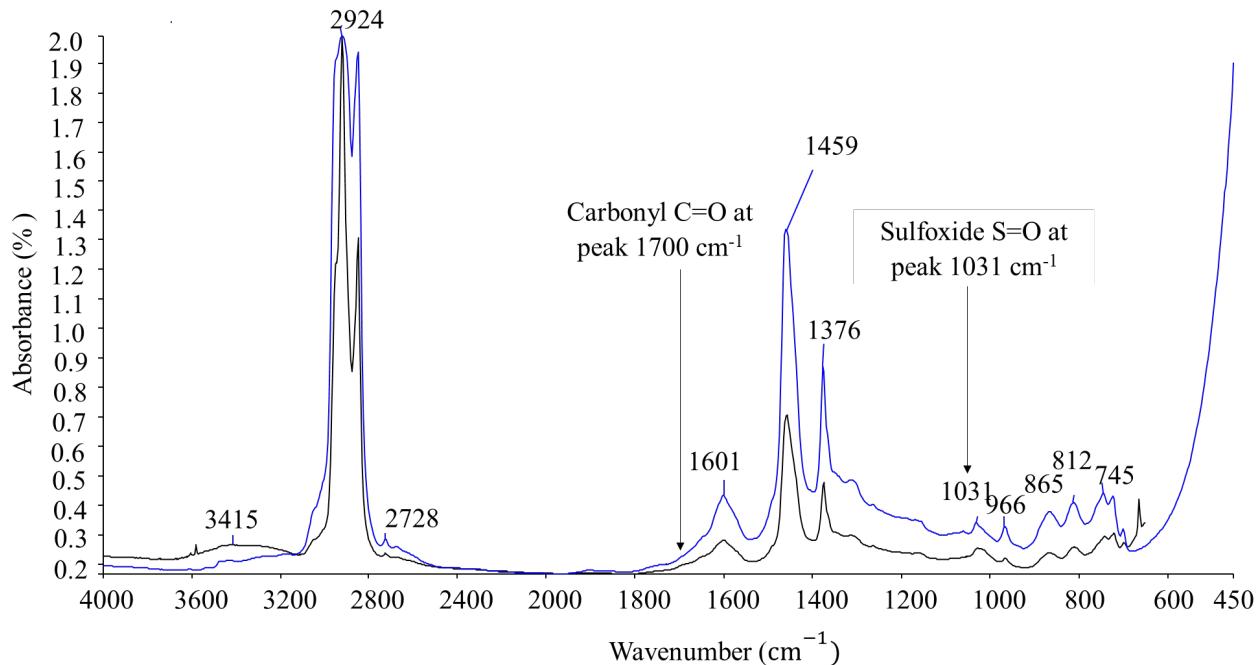
Afterwards, a thin-film sample (thickness of 0.5 mm) was laid on a newly polished salt plate sitting in a metallic frame to be scanned under the FTIR. The solvent was left to evaporate in a nitrogen environment, and then it was placed in a temperature-controlled oven at 100 °C for 10 minutes to make sure complete evaporation of the solvent is achieved. A similar ageing approach was used to characterize the ageing of bio-bitumen modified by pyrolysis oils derived from the municipal solid waste [34].

Temperature-controlled oven at 100 °C (373.15 K) under atmospheric air conditions (0.2 atm oxygen pressure) was used to age the thin-film specimens for extended periods of time up to 504 h. FTIR spectroscopy tests were carried out at different ageing periods including 0 (unaged), 12, 24, 48, 72, 96, 120, 144, 168, 336 and 504 h, using a PerkinElmer spectrum 100 spectrometer. The device was set to scan in a range of (450 to 4000) cm⁻¹, with a scanning frequency number of 32 and 4 cm⁻¹ resolution. Background scanning was made prior to samples' scanning. The tests were run in duplicates to ensure repeatability of results. The repeatability was measured by the percent of error with a maximum value of 11%. The isokinetic temperature of bitumen binders is recorded to be 100 °C [1]. At this temperature, the oxidation activation energies are independent on the binder source. For this reason, the oven-ageing temperature was selected to be 100 °C to eliminate the binder source effects. Additionally, at 100 °C temperature, the bitumen suffers detectable ageing at short time periods since the temperature lies within the pressurized ageing vessel (PAV) testing temperature range (90-110 °C) [35].

208 **3. Data Analysis and Result Discussions**

209 **3.1 FTIR Data Analysis of AACs**

210 Figure 1 shows the FTIR results for two replicates of unaged control bitumen samples. There is a
 211 considerable difference between the two replicates at the carbonyl and sulfoxide functional groups
 212 areas (peaks 1700 cm^{-1} and 1031 cm^{-1} , respectively), even though they were prepared at identical
 213 conditions from the same binder source. This suggests the area between two valleys (the range
 214 between 1820 to 1650 cm^{-1} and 1080 to 980 cm^{-1} , for carbonyl and sulfoxide, respectively) are not
 215 representative measures for oxidation. Those two functional groups are particularly important for
 216 developing the prediction models of long-term oxidative ageing of bitumen binders and asphalt
 217 mixtures, which currently make use of plain parameters of carbonyl area (*CA*) and sulfoxide area
 218 (*SA*). This can cause serious errors in predicting the oxidative ageing based on those parameters
 219 alone. Similar variations were obtained for the AAC-modified duplicate samples, and this variation
 220 is mainly attributed to the differences in sample film thickness which causes divergence in
 221 absorption percentages, and partially due to the heterogeneity nature of the bitumen. Such
 222 variations cannot necessarily be solved by making several trials then measure the mean areas under
 223 the peak, in fact by doing so, the results will be rather random since the film thickness is more
 224 visible on the wave's lengths than the actual functional group. This leads further to the aim of this
 225 research, which is to find a reliable measure for the oxidative ageing of the modified bitumen by
 226 AACs.



227
 228 **FIGURE 1 FTIR spectrum of control bitumen samples (no additives) at a complete wave**
 229 **range of (450-4000) cm^{-1} .**

230 Figure 2 shows the FTIR spectrum for the control bitumen sample after different oven
 231 ageing periods. The figure shows no peaks in the carbonyl range (1820 to 1650 cm^{-1}) for the unaged
 232 bitumen. This suggests that considering a constant range (1820 to 1650 cm^{-1}) for the carbonyl area
 233 can cause errors or obtain negative values for *CA* for that spectrum without a peak in the carbonyl

range. The current study examined each absorption curve individually to set appropriate ranges for both *CA* and *SA* by connecting the left valley's bottom of the peak to the right valley's bottom.

Noticeable and gradual growth in carbonyl area along the ageing period can be seen in Figure 3, in contrast to the sulfoxide, which stabilized with time after an initial sudden growth at the first 24 hours of ageing. This behaviour matches with the findings of Zhao et al. (2010) and Ma et al. (2012) who used Rolling Thin Film Oven (RTFO) ageing protocol in which they studied the ageing properties of SBS polymer modified bitumen [4, 13].

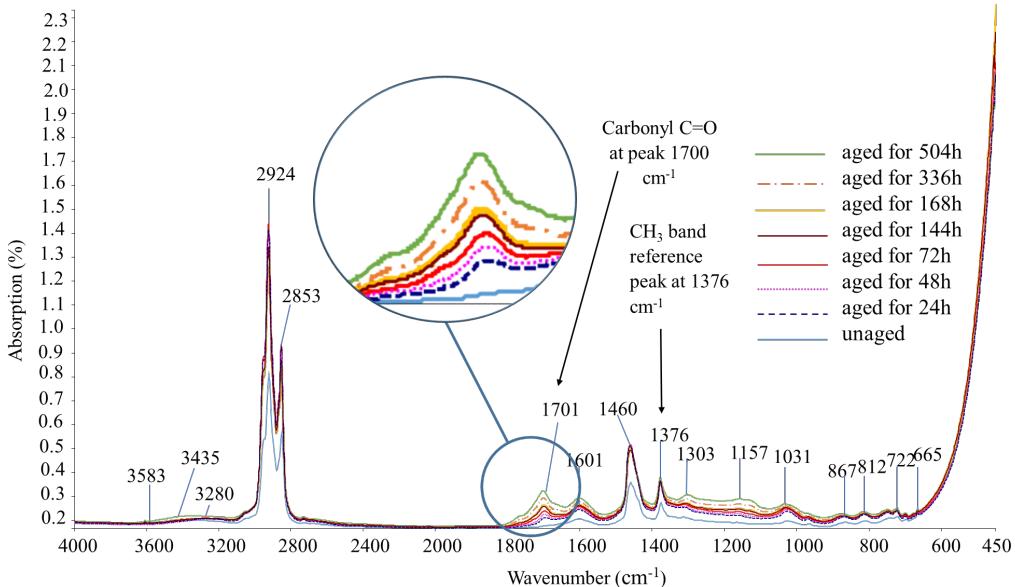


FIGURE 2 FTIR spectrum of control bitumen sample (no additives) for different ages, at a complete waves range of (450- 4000) cm⁻¹.

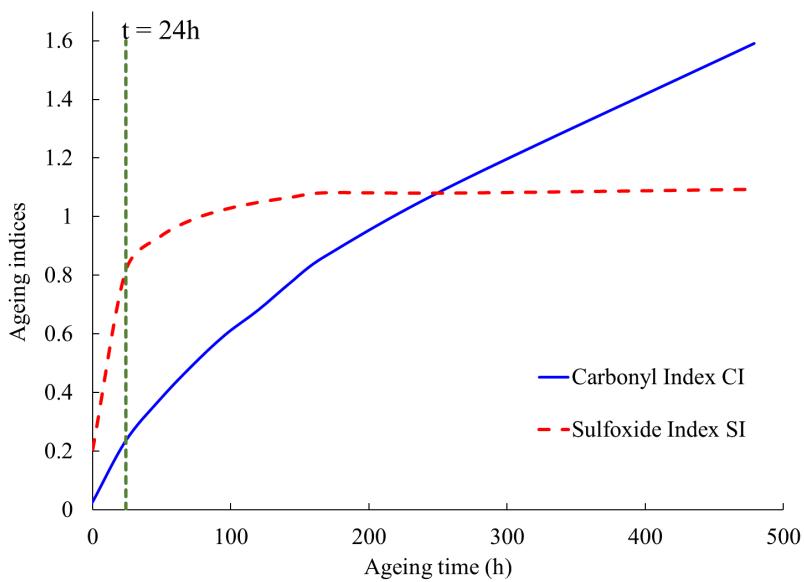


FIGURE 3 Carbonyl and sulfoxide indices of control bitumen sample (no additives) measured with respect to the reference peak 1377 cm⁻¹ versus the oven ageing time.

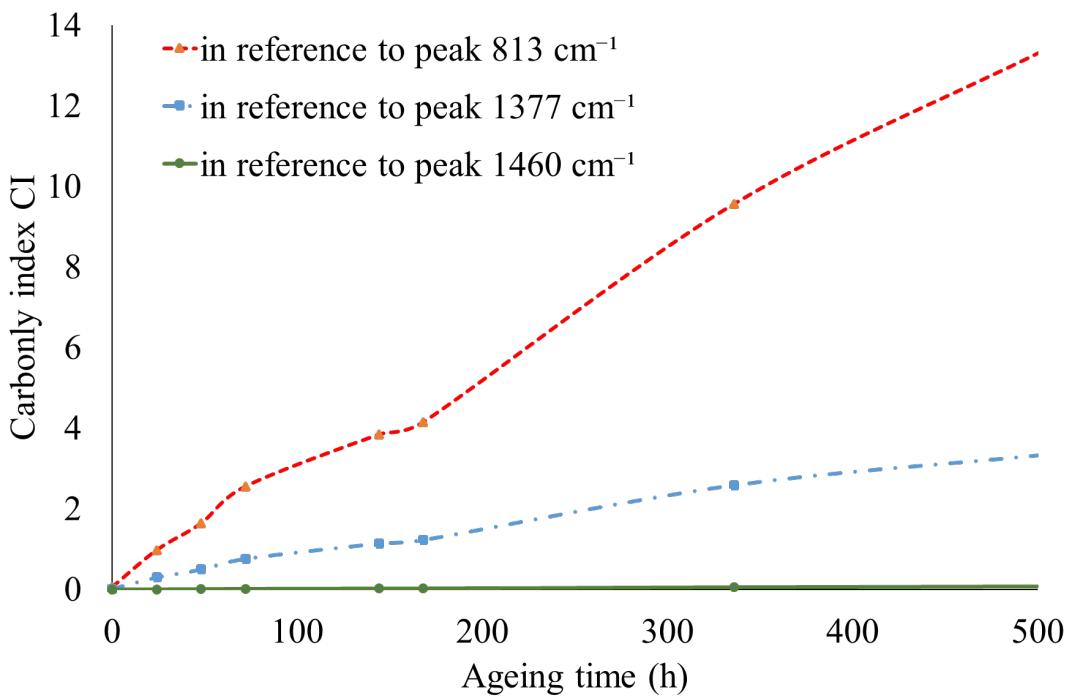
247 To eliminate the sample film thickness effect on the ageing parameters (*CA* and *SA*)
 248 between samples, a reference peak was used to determine the relative increase of the ageing
 249 (compared to the unaged bitumen) due to the extension of the ageing period. The *Reference peaks*
 250 can be determined from Figure 2 as those functional groups' peaks that maintained their areas
 251 unchanged during ageing.

252 From the bitumen spectrum's footprint, some peaks can be distinguished, peak 2924 cm^{-1}
 253 has shoulders of 2853 cm^{-1} and 2953 cm^{-1} , these belong to alkyl C-H functional group. Although
 254 this range was used by several studies as a reference peak [13], it can cause errors due to its high
 255 absorption percentage compared to the carbonyl. Whereas peak 1601cm^{-1} is shouldered with
 256 carbonyl, therefore it is an inaccurate measure to use.

257 Other recognizable peaks are 1460 cm^{-1} , 1377 cm^{-1} and 813 cm^{-1} representing CH_2 alkanes,
 258 CH_3 alkanes and aromatics, respectively. Carbonyl index (*CI*) is developed based on Equation (1)
 259 with respect to these three different peaks. Patterns of the carbonyl index obtained using those
 260 three reference peaks were compared in Figure 4.

$$261 \quad CI = \frac{CA}{\text{Area under reference peak}} \quad (1)$$

262 Where, *CI* is carbonyl index that is the ratio of the carbonyl area (area under peak 1700 cm^{-1}) to
 263 the area under a reference peak at the same ageing period, measured in the same unit of absorption
 264 from the FTIR tests.



265

266 **FIGURE 4 Carbonyl index at different ageing periods for bitumen without additives when**
 267 **different reference peaks were used.**

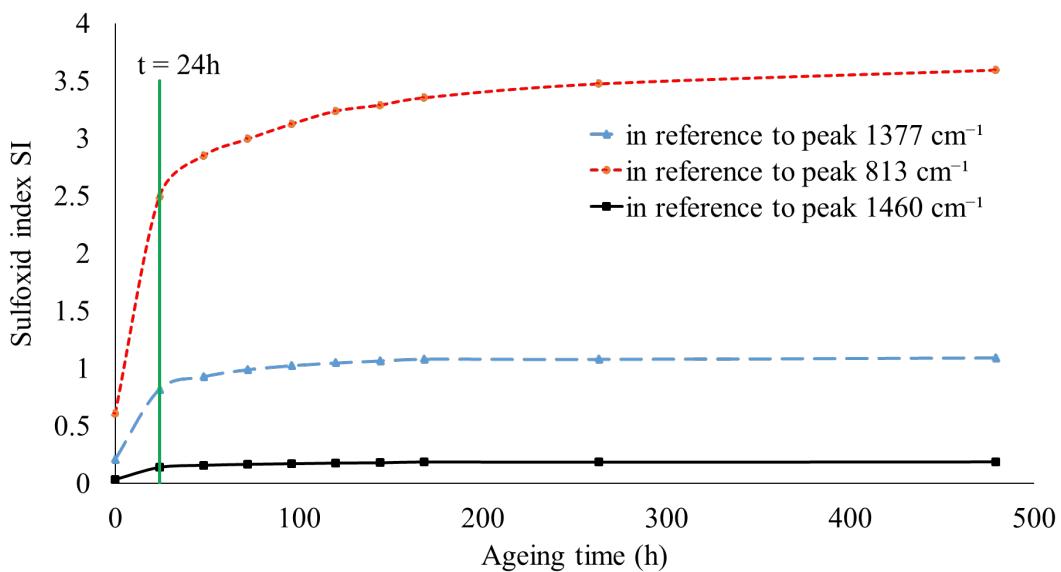
268 Peak 1377 cm^{-1} has been chosen as a reference due to its relatively compatible height
 269 compared to the carbonyl peak, which causes fewer errors. More importantly, it maintained a
 270 consistently stable condition when bitumen was mixed with other additives. On the other hand,
 271 1460 cm^{-1} peak seemed to be less stable during the initial ageing period particularly with additives
 272 containing aromatics, such as Irganox acid, Irganox 1076, MD1024 and calcium hydroxide.
 273 Finally, 813 cm^{-1} peaks may interfere with the polymer modifiers functional groups, thus was not
 274 selected.

275 Figure 5 shows the sulfoxide index (SI) (ratio of the area under peak 1031 cm^{-1} to the area
 276 under the reference peak at the same ageing period, refer to Equation (2)) of the control sample
 277 which has no additives. SI had the same pattern of CI at different reference peaks, but unlike
 278 carbonyl, sulfoxide's growth rate started to decline after 24 hours of ageing at $100\text{ }^{\circ}\text{C}$.

279

$$SI = \frac{SA}{\text{Area under reference peak}} \quad (2)$$

280 No attempts were made here to compare ageing conditions in the laboratory to that in the
 281 field, since there is an extensive literature addressing this issue [36-39], and it is not the purpose
 282 here to investigate the quantitative issues of oxidative ageing but rather the quality of these
 283 measurements. Therefore, this research will focus on carbonyl growth alone without tracking the
 284 sulfoxide, but the findings can be generalized for both of the oxidation products.



285

286 **FIGURE 5 Sulfoxide index at different ageing periods for bitumen without additives**
 287 **(control sample).**

288 The initial values of CI (shown in Figure 4) prior to oven ageing were approximately zero
 289 due to the lack of carbonyl functional groups in the virgin bitumen. However, this is not necessarily
 290 the case for different bitumen sources. It is also very common to encounter carbonyl and sulfoxide
 291 in the virgin bitumen due to the ageing in the early processing, refining, and mixing operations
 292 [40] (refer to $t=0$ in Figure 5). Therefore, for comparison purposes, CI or SI plotted against ageing

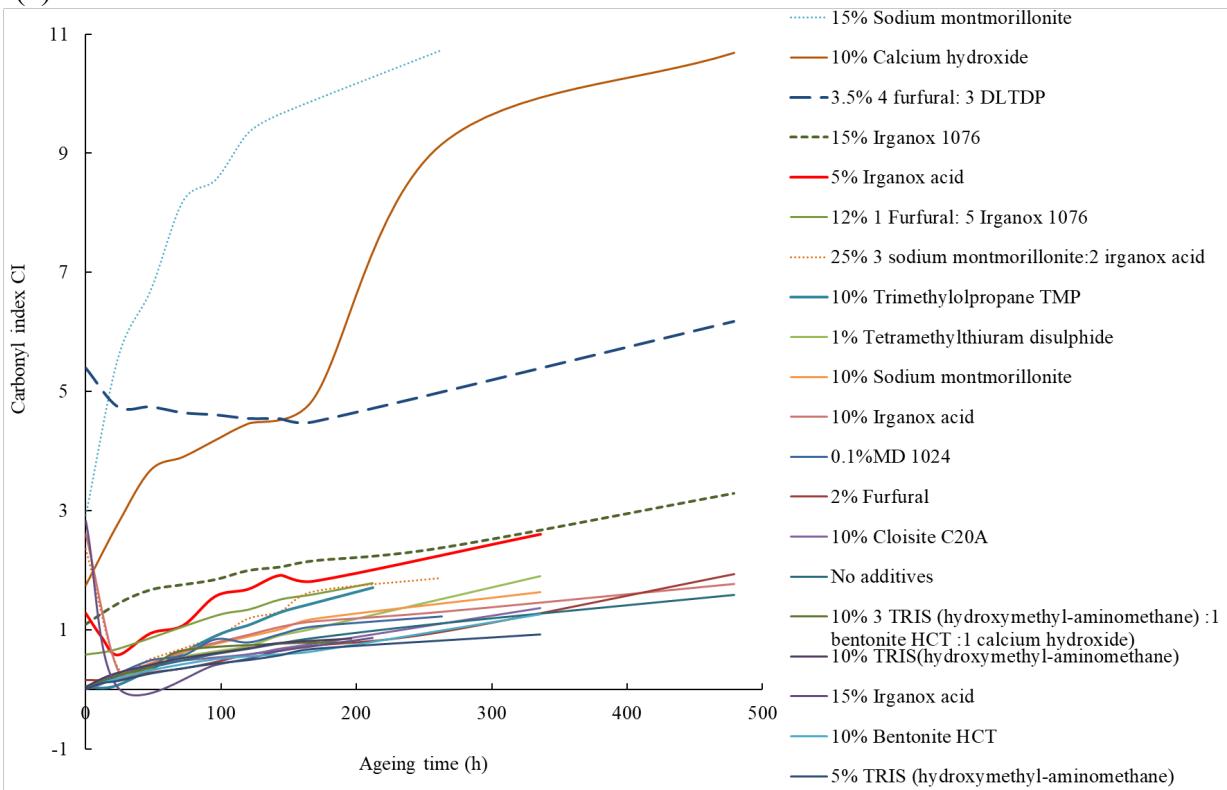
293 time may not serve the purpose of comparing the anti-oxidation performance of the materials with
294 different initial conditions.

295 Likewise, the carbonyl index plotted against ageing time for the AAC-modified bitumen
296 cannot be utilized as an AAC selection criterion due to the inherent existence of the carbonyl
297 functional group in these compounds. Figure 6 shows the *CI* for the AAC-modified and
298 unmodified bitumen versus oven-ageing time at 100 °C. It is clear that the *CI* values have different
299 initial values and can decrease when AAC-modified bitumen ages. Thus, the *CI* cannot be directly
300 used for comparing the AAC-modified bitumen in terms of its anti-ageing performance.

301 To overcome the drift at the initial values of the *CI*, another term was adopted to compare
302 the efficiency of AACs, called the Normalized Carbonyl Index *NCI*, which is presented as the ratio
303 of the difference between carbonyl index at any ageing time *CI_t* and that before ageing *CI₀* to the
304 carbonyl index at time zero *CI₀* (Equation 3).

$$305 \quad NCI = \frac{CI_t - CI_0}{CI_0} \quad (3)$$

306 According to Equation (3), the value of *NCI* will start from zero when the ageing period is
307 zero, regardless of the chemical composition of the bitumen-additive admixture. The *NCI* can drop
308 later with ageing to become a negative value. This has been observed for the AAC-modified
309 bitumen specimens that contained carbonyl in their chemical composition prior to any ageing.
310 During ageing, this carbonyl decrease led to a drop in *NCI*, causing negative values, as shown in
311 Figure 6(b).



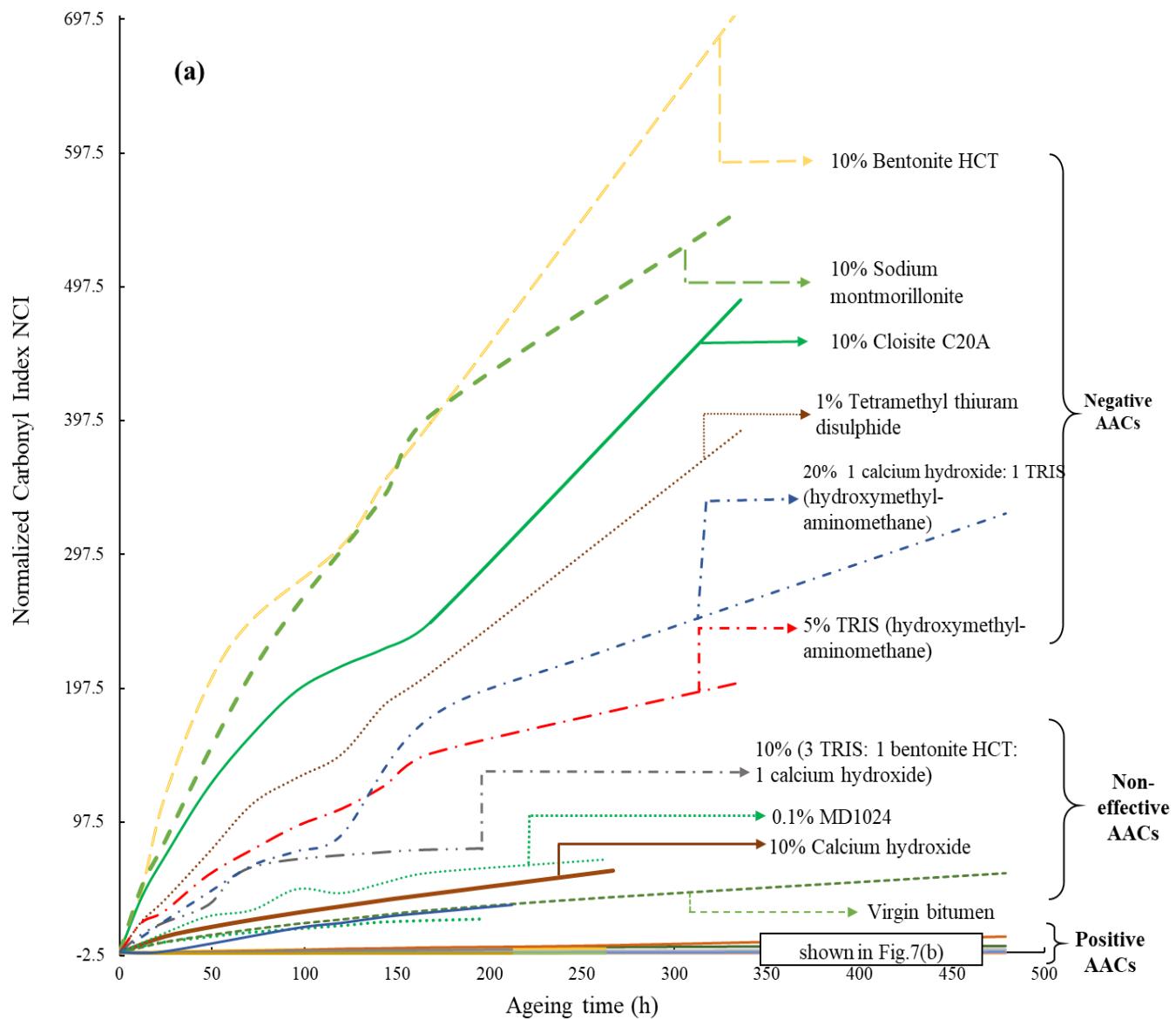
312
313 **FIGURE 6 Carbonyl index against oven ageing time at 100°C for AAC-modified and**
314 **unmodified (control) samples measured by reference peak 1377 cm⁻¹.**

315 It can be seen from Figures 7 (a and b) which show the *NCI* against ageing time for all
316 samples, AAC can be divided (according to their anti-ageing effectiveness) into three categories,
317 namely 1) negative AACs which accelerate the ageing of bitumen, 2) non-effective AACs which
318 do not alter the ageing of bitumen significantly and 3) positive AACs which inhibit the formation
319 of carbonyl and reduce bitumen ageing.

320 Negative additives included sodium montmorillonite, low concentrations of TRIS and the
321 combination of TRIS with nanomaterials and calcium hydroxide. Generally, all the tested
322 nanomaterials didn't show any positive impacts (inhibiting the carbonyl formation) on the binder's
323 ageing. This may be attributed to the samples' nature. Since the bitumen samples were thin-film
324 slides and they are assumed to be completely exposed to air oxidation. Therefore, the
325 nanomaterials added to the bitumen will introduce no resistance to the oxygen diffusion in the
326 bitumen sample. Therefore, the additives that work towards reducing diffusion will have no effect
327 on such type of samples, and further study is needed on the anti-ageing mechanism of the
328 nanomaterials on the bitumen.

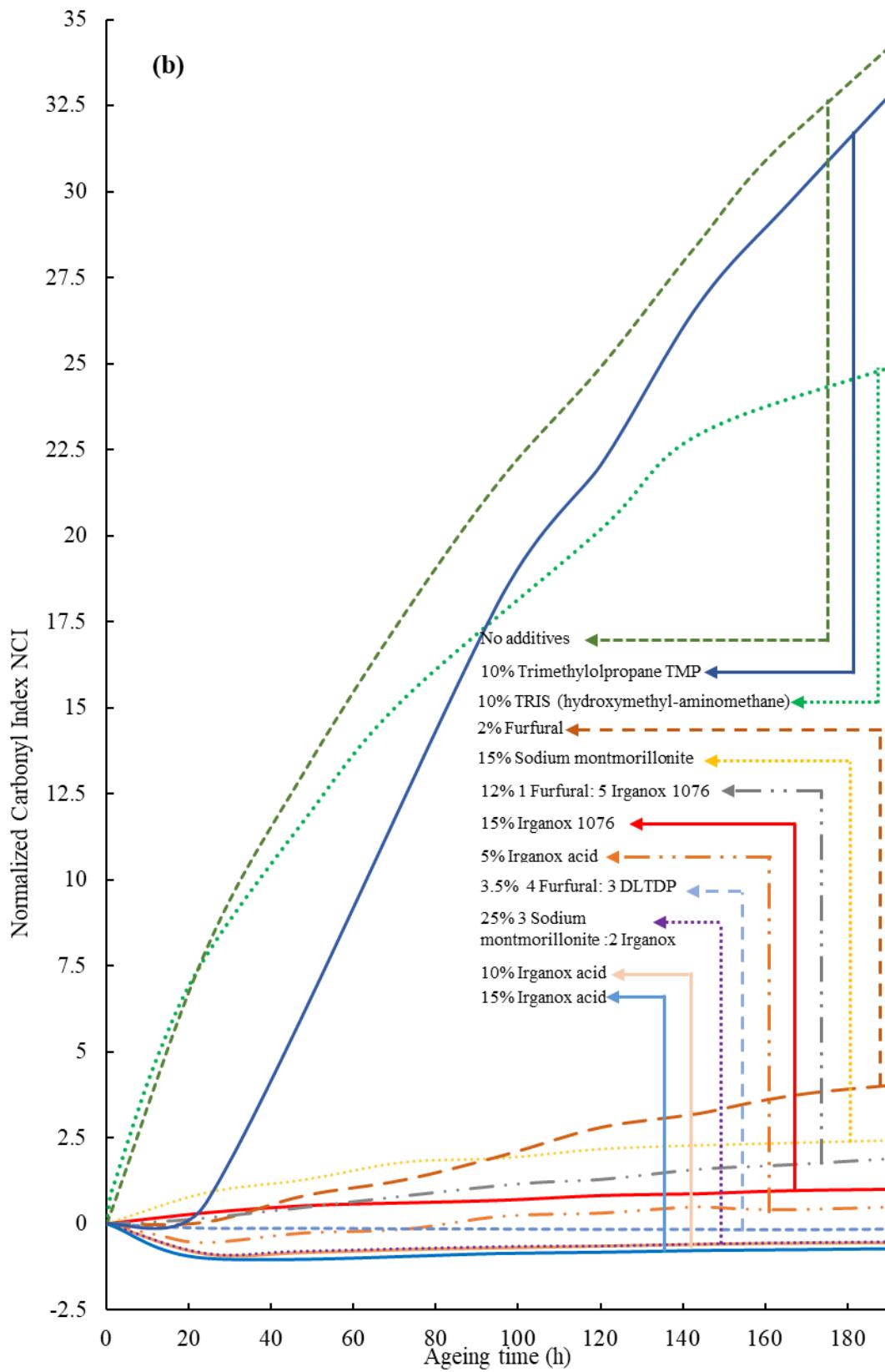
329 Non-effective additives that didn't contribute to ageing included TMP additive, calcium
330 hydroxide and 10% concentration of TRIS, as the bitumen modified with those additives show
331 comparable growth of the *NCI* to the control virgin bitumen. However, increasing TRIS
332 concentration in the binder from 5 to 10% led to a minor decrease in ageing progression.

333 Positive additives that proved effective in reducing bitumen's ageing include Irganox 1076,
334 furfural, DLTDP with furfural combination and 10% and 15% of Irganox acid, which are shown
335 as the overlapped curved at the bottom of Figure 7(a) that are separately illustrated in Figure 7(b).
336 The bitumen modified by those additives have shown a significant reduction in the growth of the
337 *NCI* compared to that of the control virgin bitumen.



338

339



341 **FIGURE 7 Normalized carbonyl index against ageing time at 100 °C for AAC-modified**
 342 **and unmodified (control) samples measured by reference peak 1377 cm⁻¹ (a) all samples,**
 343 **(b) bitumen samples with positive anti-ageing AACs.**

344 3.2 Ageing Kinetics Analysis of the AAC-modified Bitumen

345 To further compare the anti-ageing performance of different AACs, the ageing kinetics of the
 346 AAC-modified bitumen samples were investigated using Equations (4) and (5) which model the
 347 fast-term and constant-term ageing, respectively [39]. The following models have been formulated
 348 to determine the ageing rate of the binders from different sources.

349
$$\frac{dCA}{dt} = Mk_f e^{-k_f t} \quad (4)$$

350
$$\frac{\partial CA}{\partial t} = A_c P^\alpha e^{\frac{-E_c}{RT}} \quad (5)$$

351 and,

352
$$k_f = A_f P^\alpha e^{\frac{-E_f}{RT}} \quad (6)$$

353 Where, CA is the carbonyl area; M is the limiting amount of the carbonyl formation due to the first
 354 order reaction following the hot mix production; A_f and A_c are frequency factors for fast-term and
 355 constant-term ageing, and they are bitumen-type dependent, measured by (1/day) units; P is the
 356 partial pressure of oxygen, taken to be 0.2 atm in the atmosphere since the experimental ageing
 357 conditions are in the atmosphere and didn't cause any pressure changes; α is the reaction order due
 358 to oxygen partial pressure; E_f and E_c are the activation energies for both ageing stages; T is the
 359 absolute temperature measured in K; t is ageing time (days); R is the universal gas constant which
 360 equals 8.314 J/K/mol and k_f is a reaction constant which depends on the pressure and temperature.

361 It can be observed from the ageing pattern of the control sample, the carbonyl formation is
 362 time-dependent throughout the entire ageing period, suggesting the binder is still in the fast-term
 363 ageing stage. Therefore, Equation (4) can be formulated by replacing CA with NCI to model the
 364 fast-term ageing of the bitumen modified with the AACs, resulting in Equation (7).

365
$$\frac{dNCI}{dt} = Mk_f e^{-k_f t} \quad (7)$$

366 Since the pressure was kept constant during the testing and α is bitumen-type dependent, then the
 367 Equation can be simplified by using the term A'_f (using an earlier approach for modelling
 368 oxidation, proposed by Lau et al. (1992) [41]).

369
$$A'_f = A_f P^\alpha \quad (8)$$

370 According to Jin (2012) [42], the fast-term frequency factor and activation energy are related by,

371
$$A'_f = 2.031 e^{0.3076 E_f} \quad (9)$$

372 Substituting Equation (9) into Equation (6) produces Equation (10), which can be used along with
 373 Equation (7) to determine the rate of NCI ;

374
$$k_f = 2.031 e^{0.3076 E_f} e^{-\frac{E_f}{RT}} \quad (10)$$

375 By applying the curve-fitting approach to the testing data using Equation (11), k_f and M can be
 376 obtained. Then by employing Equation (10), E_f will be determined.

377
$$NCI = M(a - e^{-k_f t}) \quad (11)$$

378 Where a equal 1 ± 0.02 , is the initial error resulting from the curve fitting process. M value is a
 379 binder source-related parameter [43]. Since one type of bitumen is used in this study, therefore the
 380 variations in M and k_f values are a result of the AACs inclusion.

381 It is noteworthy to mention that there is a significant difference in the activation energy E_f
 382 , M and k_f compared with that in the literature. This is due to that the oxidation quantity is measured
 383 by the normalized carbonyl index (NCI) instead of the carbonyl area. Therefore, the model
 384 coefficients vary accordingly. However, the relationship between the A_f and E_f values in Equation
 385 (9) is still valid since A_f is not related to the oxidation rate but rather to the oxygen partial pressure
 386 and binder source, particularly since the oxygen pressure is 0.2 atm which is the same as the
 387 conditions attained in literature.

388 Table 3 shows the modified oxidation kinetics parameters which were obtained by fitting
 389 the data into Arrhenius exponential expression in Equation (11), with a coefficient of
 390 determination R^2 of more than 0.95 for most of the samples excluding bitumen sample with 2%
 391 furfural and 1.2% DLTDP (which achieved R^2 of 0.77) due to its low oxidation rate. In addition,
 392 two samples (10% and 15% of Irganox acid modified bitumen) didn't show any signs of oxidation
 393 initiation thus didn't fit to the Arrhenius model for fast-term ageing (Equation 11). On the contrary,
 394 the carbonyl amount characterised by NCI remained unchanged along the entire ageing period,
 395 which indicates an excellent anti-ageing performance of the AACs.

396

397 **TABLE 3 Oxidation kinetics model coefficients for all tested AAC modified bitumen samples**

AAC modified bitumen samples	M (1/day)	k_f (1/day)	E_f (kJ/mol)	R^2	Effectiveness of the AACs
10% Trimethylolpropane TMP	95.576	0.98	49	0.9888	Negative AACs (increase the ageing of virgin bitumen)
10% (3:1:1) TRIS (hydroxymethyl- aminomethane): bentonite HCT: calcium hydroxide	85.106	0.351	119	0.9830	
15% Sodium montmorillonite	2.635	0.336	122	0.9940	
10% TRIS (hydroxymethyl- aminomethane)	27.448	0.286	133	0.9953	
5% Irganox acid	2.101	0.154	175	0.9761	
10% Sodium montmorillonite	674.762	0.124	190	0.9988	
5% TRIS (hydroxymethyl- aminomethane)	246.71	0.116	194	0.9961	
10% Calcium hydroxide	92.539	0.088	213	0.9453	Non-effective AACs (not alter the ageing of virgin bitumen)
Virgin bitumen with no additives (control)	72.48	0.076	222	0.9962	
12%(1:5) furfural: Irganox 1076	4.413	0.075	223	0.9953	
20% (1:1) calcium hydroxide: TRIS (hydroxymethyl-aminomethane)	542.013	0.049	253	0.9839	
10% Cloisite C20A	1402.732	0.046	257	0.9833	
15% Irganox 1076	3.023	0.045	258	0.9710	

0.1% MD 1024	238.879	0.033	278	0.9703	
10% Bentonite HCT	1275.739	0.031	283	0.9803	
1% Tetramethylthiuram disulphide	1391.963	0.023	305	0.9972	
2% Furfural	1214.963	0.001	567	0.9914	
25% 3:2) sodium montmorillonite: Irganox acid	216.349	0.001	596	0.9653	
3.5% (4:3) furfural: DLTDP	132.086	0.001	664	0.7672*	Positive AACs (reduce the ageing of virgin bitumen)
10% Irganox acid*	--	--	--	--	
15% Irganox acid*	--	--	--	--	

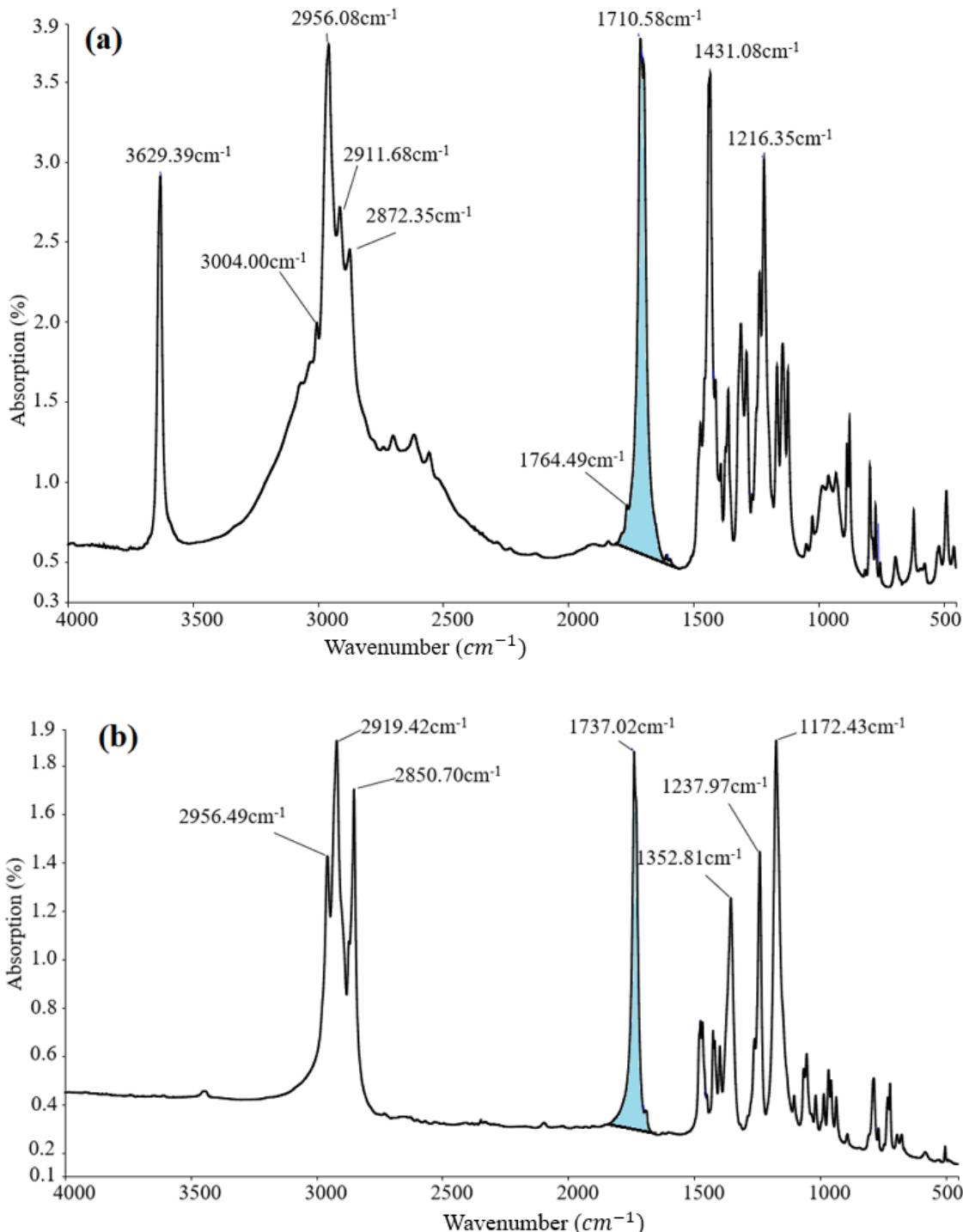
*No significant oxidation was observed for those AACs modified bitumen samples, thus the ageing kinetics models (in Equation 11) was not fitted or produced a low R^2 . This observation indicates an excellent anti-ageing effect of the AACs on the bitumen.

398

399 Samples in Table 3 are arranged according to their fast-term activation energy in
400 ascending order. Samples of small activation energies tend to have fast oxidation, while those
401 with greater oxidation energies are expected to have better anti-ageing performance, namely the
402 later develop fewer carbonyl compounds under the accelerated ageing conditions. Samples like
403 10% and 15% Irganox acid that performed superiorly compared to others didn't fit to the
404 modelled equation because they didn't show any signs of ageing under 100 °C oven temperature
405 conditions for a thin-film binder at ambient atmospheric pressure. Based on this observation, it is
406 recommended to use activation energy to group the AACs in terms of their effectiveness in anti-
407 ageing performance. Those AACs with activation energy below 200 kJ/mol are negative AACs
408 as they will increase the ageing (quantified by the formation of carbonyl compounds) of the
409 virgin bitumen and those above 300 kJ/mol are positive AACs as they will decrease the ageing of
410 the virgin bitumen. Those between 200 and 300 kJ/mol are non-effective AACs since they do not
411 alter the ageing of the virgin bitumen.

412 There was a great variation in M values for the tested specimens which can be attributed to
413 the chemical composition of the additives. Furthermore, some of the AACs contained carbonyl
414 functional groups in their primary form prior to ageing. Figure 8 (a and b) shows the chemical
415 composition of Irganox acid and DLTDP, where a well-defined peak for the carbonyl functional
416 groups can be observed at waves peak from 1710 to 1737 cm⁻¹. This AAC-induced carbonyl
417 functional groups led to the negative values on the initial portion of the NCI vs ageing time curves,
418 especially during the ageing period less than one day ($t = 1$ day). This portion was neglected in the
419 curve-fitting process since it makes nonsense to fit the negative NCI values to the suggested ageing
420 kinetics model in Equation (11). Logically, this affects the M values. Therefore, M values cannot
421 be used as an ageing indicator for the screening purposes of variable AAC additives to evaluate
422 the anti-ageing performance of the AAC modified bitumen.

423



424

425

426 **FIGURE 8 FTIR spectrum of (a) Irganox acid and (b) DLTDP at a complete waves range
427 of (450- 4000) cm^{-1}**

428 According to the activation energy based AAC categories obtained in Table 3, the negative
429 AACs lead to a drop in the oxidation energy of binders thus increased the oxidative ageing
430 susceptibility, which are sodium montmorillonite, Tris and Ca(OH)_2 . For the non-effective AACs,

431 it appears that adding those AAC does not alter the bitumen's ageing performance, such as a low
432 concentration of 5% by binder's weight of Irganox acid, Irganox 1076, C20A, and 0.1% MD 1024.
433 Nano-additives such as bentonite and Tetramethylthiuram disulphide are also non-effective AACs
434 which appeared to cause an insignificant increase in the activation energy.

435 The positive AACs exhibited a high anti-ageing performance by increasing the oxidation
436 energy of the bitumen and those positive AACs were those containing furfural, Irganox acid with
437 sodium montmorillonite, furfural with DLTDP, and high concentrations of Irganox acid (10% or
438 more). The exact amount of increase in E_f for Irganox acid was beyond measure at current testing
439 conditions, where bitumen with Irganox acid didn't show any signs of carbonyl development. In
440 sum, with the aid of activation energy based AAC categories and the proposed NCI as the ageing
441 characterisation parameter, it can be concluded that this NCI and activation energy based ageing
442 evaluation approach has proved convenience and effective as an initial screening method for anti-
443 ageing compounds for bitumen.

444 4. Conclusions

445 This paper investigates the effect of anti-ageing compounds (AACs) on the oxidation kinetics of
446 bitumen using FTIR tests. Up to 20 different AACs were added to bitumen binder and subjected
447 to laboratory oven ageing conditions at 100°C for a range of ageing durations. The results indicated
448 the followings:

- 449 1) A reference peak (at 1377 cm⁻¹ wavelength) was identified to determine the carbonyl index
450 to quantify the ageing in bitumen, due to its relatively comparable height to the carbonyl
451 which causes fewer errors. An advantage of using the peak at 1377 cm⁻¹ lies in that it
452 remained unchanged when bitumen was modified by different AACs and aged for different
453 durations.
- 454 2) The carbonyl content measured using carbonyl areas (CA) under the waves range (1820-
455 1650 cm⁻¹) or any other fixed ranges, was found unsuitable for the oxidative ageing
456 quantification of bitumen. It cannot be used to compare the ageing of the formulated
457 samples because of specimens' disparities in terms of bitumen sources, AAC types and
458 sample thickness. Using a fixed waves range may cause negative values for CA, especially
459 for the unaged binders or short-term aged samples. Therefore, the wave range should be
460 inspected for each sample.
- 461 3) Carbonyl index (CI, defined as the ratio of the carbonyl area to the reference peak area)
462 was ineffective in evaluating AACs' anti-ageing performance. This is due to that some
463 bitumen binders and AACs anti-oxidants contain carbonyl in their initial formulas, thus CI
464 versus ageing time for those compounds may start from a non-zero CI value, resulting in a
465 non-comparable growth rate between bitumen modified by different AACs.
- 466 4) Normalized carbonyl index (NCI) was proposed to quantify the oxidative ageing of
467 bitumen modified by AACs. It eliminates the impact of the initial carbonyl content of the
468 binders or AACs, thus manifests the carbonyl growth rate to evaluate the AACs' anti-
469 ageing performance. It was found that NCI can be used to formulate the ageing kinetics of
470 the AAC-modified bitumen, where kinetics model coefficients M, K_f and E_f values were
471 obtained using the data of NCI vs. ageing time.

- 472 5) The activation energy E_f was found to be capable of the differentiating the anti-ageing
 473 efficiency of the AACs. Those AACs with E_f below 200 kJ/mol are negative as they will
 474 increase the ageing (quantified by the formation of carbonyl compounds) of the virgin
 475 bitumen and those above 300 kJ/mol are positive AACs as they will decrease the ageing of
 476 the virgin bitumen. Those between 200 and 300 kJ/mol are non-effective AACs since they
 477 do not alter the ageing of the virgin bitumen.
- 478 6) Samples that exhibited high anti-ageing performance were those contained furfural,
 479 Irganox acid with sodium montmorillonite, furfural with DLTDP, and high concentrations
 480 of Irganox acid (10% or more).

481 This study proposed an initial screening and analysing methodology to evaluate the anti-ageing
 482 efficiency of the AACs when used in bitumen. This method can provide a primary screening of
 483 the additives' effectiveness at reducing the ageing rate of the modified bitumen, however, it should
 484 be followed by further laboratory rheological and mechanical tests on the AAC modified bitumen
 485 with different binder sources in future studies.

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